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# JEE Advanced: Paper-I (2018)

# **IMPORTANT INSTRUCTIONS**

- 1. Each question has FOUR options for correct answer(s). ONE OR MORE THAN ONE of these four option(s) is(are) correct option(s).
- For each question, choose the correct option(s) to answer the question.
- 3. Answer to each question will be evaluated according to the following marking scheme:

Full Marks: +4 If only (all) the correct option(s) is (are) chosen.

Partial Marks: +3 If all the four options are correct but ONLY three options are chosen.

Partial Marks: +2 If three or more options are correct but ONLY two options are chosen, both

of which arecorrect options.

Partial Marks: +1 If two or more options are correct but ONLY one option is chosen and it is a

correct option.

`Zero Marks: 0 If none of the options is chosen (i.e. the question is unanswered).

Negative Marks: -2 In all other cases.

4. For Example: If first, third and fourth are the ONLY three correct options for a question with second option

being an incorrect option; selecting only all the three correct options will result in +4 marks.

Selecting only two of the three correct options (e.g. the first and fourth options), without selecting any incorrectoption (second option in this case), will result in +2 marks. Selecting only one of the three correct options (eitherfirst or third or fourth option) without selecting any incorrect option (second option in this case), will result in +1marks. Selecting any incorrect option(s) (second option in this case), with or without selection of any correctoption(s) will result in -2 marks.

## PART-A: MATHEMATICS

The potential energy of a particle of mass m at a distance r from a fixed point O is given by  $V(r) = kr^2/2$ , 1. where k is a positive constant of appropriate dimensions. This particle is moving in a circular orbit of radius R about the point O. If v is the speed of the particle and L is the magnitude of its angular momentum about O, which of the following statements is (are) true?

(A) 
$$v = \sqrt{\frac{k}{2m}}R$$

(B) 
$$v = \sqrt{\frac{k}{m}}R$$

(C) 
$$L = \sqrt{m k} R^2$$

(C) 
$$L = \sqrt{m k} R^2$$
 (D)  $L = \sqrt{\frac{m k}{2}} R^2$ 

Ans. [BC]

Sol. 
$$\vec{F} = \frac{-\partial V}{\partial r}(\hat{r})$$

$$\vec{F} = -Kr(\hat{r})$$

$$F_{centripital} = \frac{mv^2}{R}$$

$$KR = \frac{mv^2}{R}$$

$$v = \sqrt{\frac{K}{m}} \ R$$

$$L = m \cdot \sqrt{\frac{K}{m}} R \cdot R$$

$$L = \sqrt{mK} \cdot R^2$$

Consider a body of mass 1.0 kg at rest at the origin at time t = 0. A force  $\vec{F} = (\alpha t \hat{i} + \beta \hat{j})$  is applied on the 2. body, where  $\alpha = 1.0 \text{ N s}^{-1}$  and  $\beta = 1.0 \text{ N}$ . The torque acting on the body about the origin at time  $t = 1.0 \text{ s is } \vec{\tau}$ . Which of the following statements is (are) true?

(A) 
$$|\vec{\tau}| = \frac{1}{3} Nm$$

- (B) The torque  $\vec{\tau}$  is in the direction of the unit vector +  $+\hat{k}$
- (C) The velocity of the body at t = 1 s is  $\vec{v} = \frac{1}{2}(\hat{i} + 2\hat{j})\text{ms}^{-1}$
- (D) The magnitude of displacement of the body at t = 1 s is  $\frac{1}{6}$ m

[AC] Ans.

$$\textbf{Sol.} \qquad \vec{\mathsf{F}} \,=\, \alpha\,t\,\hat{\mathsf{i}} \,+\,\beta\,\,\hat{\mathsf{j}}$$

$$\vec{a} = \alpha t \hat{i} + \beta \hat{j}$$

$$\vec{v} = \frac{\alpha t^2}{2} \hat{i} + \beta t \hat{j}$$

$$= + \vec{r} = \frac{\alpha t^3}{6} \hat{i} + \frac{\beta t^2}{2} \hat{j}$$

$$\vec{\tau} = \vec{r} \times \vec{F}$$

$$\vec{\tau} = \frac{\alpha t^3 \beta}{6} \hat{k} - \frac{\alpha t^3 \beta}{2} \hat{k}$$

$$\left| \left( \vec{\tau} \right)_{at} t = 1 \operatorname{sec} \right| = \frac{1}{3} N - m$$

$$\vec{v}_{\text{att=1sec}} = \frac{1}{2}\hat{i} + \hat{j}$$

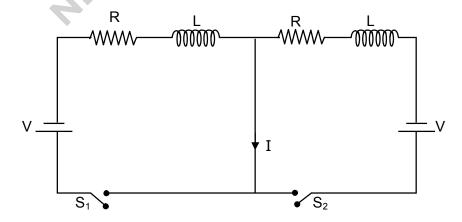
$$=\frac{1}{2}(\hat{i}+2\hat{j})m sec^{-1}$$

- 3. A uniform capillary tube of inner radius r is dipped vertically into a beaker filled with water. The water rises to a height h in the capillary tube above the water surface in the beaker. The surface tension of water is σ. The angle of contact between water and the wall of the capillary tube is θ. Ignore the mass of water in the meniscus. Which of the following statements is (are) true?
  - (A) For a given material of the capillary tube, h decreases with increase in r
  - (B) for a given material of the capillary tube, h is independent of  $\sigma$
  - (C) If this experiment is performed in a lift going up with a constant acceleration, then h decreases
  - (D) h is proportional to contact angle  $\theta$

Ans. [AC]

**Sol.** 
$$h = \frac{2\sigma \cos \theta}{\rho q_{off} r}$$

- (A) as r increases h decreases
- (C) as lift is accelerated upwards  $g_{\rm eff}$  = g + a, as a result of which h decreases.
- In the figure below, the switches  $S_1$  and  $S_2$  are closed simultaneously at t = 0 and a current starts to flow in the circuit. Both the batteries have the same magnitude of the electromotive force (emf) and the polarities are as indicated in the figure. Ignore mutual inductance between the inductors. The current I in the middle wire reaches its maximum magnitude  $I_{max}$  at time  $t = \tau$ . Which of the following statements is (are) true?



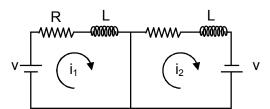
(A) 
$$I_{\text{max}} = \frac{V}{2R}$$

(B) 
$$I_{\text{max}} = \frac{V}{4R}$$

(C) 
$$\tau = \frac{L}{R} \ln 2$$

(B) 
$$I_{max} = \frac{V}{4R}$$
 (C)  $\tau = \frac{L}{R} \ln 2$  (D)  $\tau = \frac{2L}{R} \ln 2$ 

[BD] Ans.



Sol.

$$i_{max} = (i_2 - i_1)_{max}$$

$$i_2 - i_1 = \frac{v}{R} \left[ 1 - e^{-\left(\frac{R}{2L}\right)t} \right] - \frac{v}{R} \left[ 1 - e^{-\left(\frac{R}{L}\right)t} \right]$$

$$i = i_2 - i_1 = \frac{v}{R} \left[ e^{-\left(\frac{R}{L}\right)t} - e^{-\left(\frac{R}{2L}\right)t} \right]$$

...(i)

$$\mbox{for } i_{\mbox{\scriptsize max}} = \!\! \frac{d}{dt} \big( i_{\mbox{\scriptsize 2}} - i_{\mbox{\scriptsize 1}} \big) \ = \ 0 \label{eq:imax}$$

$$\left(\frac{R}{2L}\right)t \ = \ \ell n \ 2$$

$$t = \frac{2L}{R} \ell n 2$$

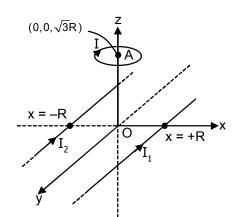
...(iii)

Putting (iii) in (ii)

$$|i_{max}| = \frac{V}{4R}$$

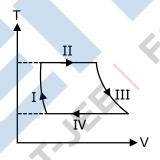
- Two infinitely long straight wires lie in the xy-plane along the lines  $x = \pm R$ . The wire located at x = +R5. carries a constant current  $I_1$  and the wire located at x = -R carries a constant current  $I_2$ . A circular loop of radius R is suspended with its centre at  $(0, 0\sqrt{3}R)$  and in a plane parallel to the xy-plane. This loop carries a constant current I in the clockwise direction as seen from above the loop. The current in the wire is taken to be positive if it is in the  $+\hat{j}$  direction. Which of the following statements regarding the magnetic field B is (are) true?
  - (A) If  $I_1 = I_2$ , then  $\overrightarrow{B}$ **cannot** be equal to zero at the origin (0, 0, 0)
  - (B) If  $I_1$ > 0 and  $I_2$ < 0, then  $\vec{B}$  can be equal to zero at the origin (0, 0, 0)
  - (C) If  $I_1 < 0$  and  $I_2 > 0$ , then  $\vec{B}$  can be equal to zero at the origin (0, 0, 0)
  - (D) If  $I_1 = I_2$ , then the z-component of the magnetic field at the centre of the loop is  $\left(-\frac{\mu_0 I}{2R}\right)$

[ABD] Ans.



Sol.

- (A) If  $I_1 = I_2$  then magnetic field due to infinitely long wires is zero but due to ring magnetic field is not zero.
- (B)  $I_1 > 0$  and  $I_2 < 0$ , then magnetic field will be in direction  $\hat{k}$  and magnetic field due to ring is in  $-\hat{k}$  direction. Hence net magmatic field at origin can be zero.
- (D) If  $I_1 = I_2$  then magnetic field due to two wires is along + x axis but due to ring it is along  $\hat{k}$  and having magnitude  $\frac{\mu_0 i}{2R}$
- 6. One mole of a monatomic ideal gas undergoes a cyclic process as shown in the figure (where V is the volume and T is the temperature). Which of the statements below is (are) true?



- (A) Process I is an isochoric process
- (B) In process II, gas absorbs heat
- (C) In process IV, gas releases heat
- (D) processes I and III are not isobaric

#### Ans. [BCD]

**Sol.** (B) Process - II is isothermal expansion

So, 
$$\Delta U = 0$$
 and  $\Delta w > 0$ 

we know that

$$\Delta Q = \Delta U + \Delta w$$

 $\Delta Q > 0$ 

(C) Process - IV is isothermal compression

$$\Delta U = 0$$
,  $\Delta w < 0$ 

$$\Delta Q = \Delta U + \Delta w$$

 $\Delta Q < 0$ 

(D) Process I and III are not isobaric as for isobaric process the line must be straight.

#### **SECTION 2**

- This section contains EIGHT (08) questions. The answer to each question is a NUMERICAL VALUE.
- For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 6.25, 7.00, -0.33, -.30, 30.27, -127.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks

+3 If ONLY the correct numerical value is

entered as answer.

Zero Marks

0 In all other cases.

7. Two vectors  $\overrightarrow{A}_{and}\overrightarrow{B}$  are defined as  $\overrightarrow{A}_{and}\overrightarrow{B}_{a$ 

Ans. [2.00]

**Sol.**  $|\vec{A} + \vec{B}| = 2a\cos\frac{\omega t}{2}$ 

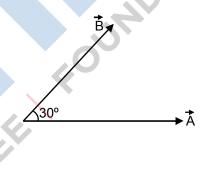
$$|\vec{A} - \vec{B}| = 2a \sin \frac{\omega t}{2}$$

from given conditions.

$$2a\frac{\cos\omega t}{2} = \sqrt{3} \cdot 2a\sin\frac{\omega t}{2}$$

$$\frac{\omega t}{2} = \frac{\pi}{6} \Rightarrow \omega t = \frac{\pi}{3}$$

$$\Rightarrow \frac{\pi}{6} \times t = \frac{\pi}{3} \Rightarrow t = 2.00 \text{ sec.}$$



8. Two men are walking along a horizontal straight line in the same direction. The man in front walks at a speed 1.0 ms<sup>-1</sup> and the man behind walks at a speed 2.0 ms<sup>-1</sup>. A third man is standing at a height 12 m above the same horizontal line such that all three men are in a vertical plane. The two walking men are blowing identical whistles which emit a sound of frequency 1430 Hz. The speed of sound in air is 330 ms<sup>-1</sup>. At the instant, when the moving men are 10 m apart, the stationary man is equidistant from them. The frequency of beats in Hz, heard by the stationary man at this instant, is \_\_\_\_\_\_.

Ans. [5.00] Sol.

$$f_{Beat} = f_A - f_B$$
 ...(1  
 $f_A = 1430 \left[ \frac{330}{330 - 2\cos\theta} \right]$ 

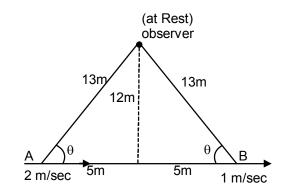
$$f_A = 1430 \left[ 1 + \frac{2\cos\theta}{330} \right] \dots (2)$$

$$f_{B} = 1430 \left[ \frac{330}{330 + 1\cos\theta} \right]$$

$$f_B = 1430 \left[ 1 - \frac{\cos \theta}{330} \right] \qquad ...(3)$$

Putting (2) and (3) in (1)

$$f_{Beat} = 5.00 Hz$$



A ring and a disc are initially at rest, side by side, at the top of an inclined plane which makes an angle 60° with the horizontal. They start to roll without slipping at the same instant of time along the shortest path. If the time difference between their reaching the ground is  $(2 - \sqrt{3}) / \sqrt{10} \, s$ , then the height of the top of the inclined plane, in metres, is \_\_\_\_\_. Take g = 10 ms<sup>-2</sup>.

Ans. [0.75]

Sol.

$$a_{c} = \frac{gsin\theta}{1 + \frac{I_{c}}{MR^{2}}}$$

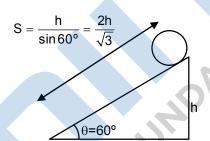
$$a_{\text{Ring}} = \frac{g \sin \theta}{2}, a_{\text{Disc}} = \frac{2g \sin \theta}{3}$$

$$t_1 = \sqrt{\frac{2s}{a_{Ring}}} = \sqrt{\frac{16h}{3g}}$$

$$t_2 = \sqrt{\frac{2s}{a_{\text{Disc}}}} = \sqrt{\frac{4h}{q}}$$

$$\sqrt{\frac{16h}{g}} - \sqrt{\frac{4h}{g}} = \frac{\left(2 - \sqrt{3}\right)}{\sqrt{10}}$$

h = 0.75 meter



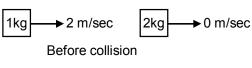
10. A spring-block system is resting on a frictionless floor as shown in the figure. The spring constant is 2.0 N m<sup>-1</sup> and the mass of the block is 2.0 kg. Ignore the mass of the spring. Initially the spring is in an unstretched condition. Another block of mass 1.0 kg moving with a speed of 2.0 ms<sup>-1</sup> collides elastically with the first block. The collision is such that the 2.0 kg block does not hit the wall. The distance, in metres, between the two blocks when the spring returns to its unstretched position for the first time after the collision is



Ans. [2.09]



Sol.



1kg

After collision

from momentum conservation  $(1 \times 2) + (2 \times 0) = (1 \times v_1) + (2 \times v_2)$  .... (1) as collision is elastic, so KE is conserved.

$$\frac{1}{2} \times 1(2)^{2} + \frac{1}{2} \times 2(0)^{2} = \frac{1}{2} \times 1 \times (v_{1})^{2} + \frac{1}{2} \times 2 \times (v_{2})^{2} \dots (2)$$

from (1) and (2)

$$v_1 = -\frac{2}{3}m / sec$$

$$v_2 = \frac{4}{3} m / sec$$

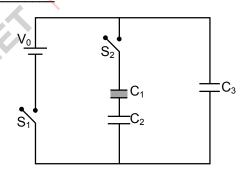
$$T_{\text{oscillation}} = 2\pi \sqrt{\frac{m}{k}} = 2\pi \sqrt{\frac{2}{2}} = 2\pi \, \text{sec}$$

When spring returns to its unstretched position for first time, time spent =  $\frac{T}{2}$  =  $\pi$  sec

Distance covered  $=\frac{2}{3} \times \pi = 2.093 \text{m}$ 

d = 2.09 meter

11. Three identical capacitors  $C_1$ ,  $C_2$  and  $C_3$  have a capacitance of 1.0  $\mu$ F each and they are uncharged initially. They are connected in a circuit as shown in the figure and  $C_1$  is then filled completely with a dielectric material of relative permittivity  $\in_r$ . The cell electromotive force (emf)  $V_0 = 8 \text{ V}$ . First the switch  $S_1$  is closed while the switch  $S_2$  is kept open. When the capacitor  $C_3$  is fully charged,  $S_1$  is opened and  $S_2$  is closed simultaneously. When all the capacitors reach equilibrium, the charge on  $C_3$  is found to be 5  $\mu$ C. The value of  $\in_r =$ \_\_\_\_\_\_.



Ans. [1.50]

Sol. When  $S_1$  (closed) and  $S_2$  (open)

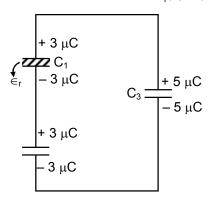
Charge on  $C_3 = 8 \mu C$ 

and  $\mathrm{C_1}$  and  $\mathrm{C_2}$  with be uncharged.

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Now when

 $S_1$ (open) and  $S_2$  (closed)

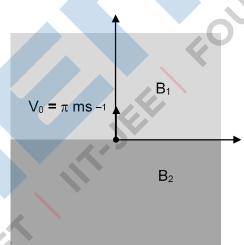


New charge on  $\rm C_3$  = 5  $\rm \mu C$  so charge on  $\rm C_1$  and  $\rm C_2$  = 3  $\rm \mu C$ 

Applying loop rule for the above circuit

$$\frac{5}{1} = \frac{3}{\epsilon_r} + \frac{3}{1} \Longrightarrow \epsilon_r = 1.50$$

12. In the xy-plane, the region y > 0 has a uniform magnetic field  $B_1\hat{k}$  and the region y < 0 has another uniform magnetic field  $B_2\hat{k}$ . A positively charged particle is projected from the origin along the positive y-axis with speed  $v_0 = \pi \text{ ms}^{-1}$  at t = 0, as shown in the figure. Neglect gravity in this problem. Let t = T be the time when the particle crosses the x-axis from below for the first time. If  $B_2 = 4B_1$ , the average speed of the particle, in ms<sup>-1</sup>, along the x-axis in the time interval T is \_\_\_\_\_\_.



Ans. [2.00]

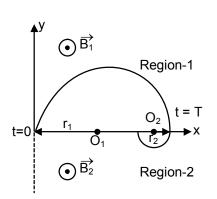
**Sol.** Radius of curvature R = 
$$\frac{m v}{qB}$$

$$r_1 = \frac{mv}{qB_1}$$

$$r_2 = \frac{m v}{q B_2}$$

$$B_2 = 4B_1$$

So 
$$r_1 = 4r_2$$



time spent in 
$$B_1 = \frac{\pi m}{qB_1} = 4 t$$

time spent in 
$$B_2 = \frac{\pi m}{qB_2} = \frac{\pi m}{4qB_1} = t$$

$$(v_{average})_{along \ x - axis} = \frac{totaldistance}{total time}$$

$$=\frac{2r_1+2r_2}{5t}=\frac{10r_2}{5t}=\frac{2r_2}{t}$$

$$v_{\text{average}} = \frac{\frac{2 \times mv}{4qB_1}}{\frac{\pi m}{4qB_1}} = \frac{2mv}{\pi m} = \frac{2v}{\pi} = 2.000 \,\text{m/sec}$$

 $v_{average} = 2.00 \text{ m/sec}$ 

Sunlight of intensity 1.3 kW m<sup>-2</sup> is incident normally on a thin convex lens of focal length 20 cm. Ignore 13. the energy loss of light due to the lens and assume that the lens aperture size is much smaller than its focal length. The average intensity of light, in kW m<sup>-2</sup>, at a distance 22 cm from the lens on the other side is \_\_\_\_\_

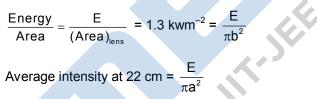
b

Ans. [130]

**Sol.** 
$$\frac{b}{20} = \frac{a}{2} \Rightarrow \frac{a}{b} = \frac{1}{10}$$

let energy falling on lens be E so accoding to given condition

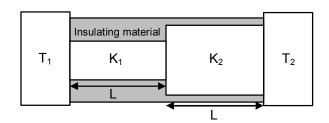
$$\frac{\text{Energy}}{\text{Area}} = \frac{\text{E}}{(\text{Area})_{\text{lens}}} = 1.3 \text{ kwm}^{-2} = \frac{\text{E}}{\pi b^2}$$



$$= \frac{1.3 \frac{\text{kw}}{\text{m}^2} \times \pi b^2}{\pi a^2} = 1.3 \times \left(\frac{b}{a}\right)^2 = 1.3 \times (10)^2$$

$$= 130 \text{ kw/m}^2$$

14. Two conducting cylinders of equal length but different radii are connected in series between two heat baths kept at temperatures  $T_1$  = 300 K and  $T_2$  = 100 K.as shown in the figure. The radius of the bigger cylinder is twice that of the smaller one and the thermal conductivities of the materials of the smaller and the larger cylinders are  $K_1$  and  $K_2$  respectively. If the temperature at the junction of the two cylinders in the steady state is 200 K, then  $K_1/K_2 =$ \_\_\_\_\_



Ans. [4.00]

300 k i<sub>1</sub> 200 k i<sub>2</sub> R<sub>1</sub> R<sub>2</sub> 100 l

Sol.

$$R_1 = \frac{L}{K_1 \pi R_1^2}$$
 ...(1)

$$R_2 = \frac{L}{K_2 \pi R_2^2}$$
 ...(2)

$$\frac{R_2}{R_1} = 2$$
 ...(3)

at steady state

$$i_1 = i_2$$

$$\frac{300 - 200}{L \over K_1 \pi R_1^2} = \frac{200 - 100}{L \over K_2 \pi R_2^2}$$

$$\frac{K_1}{K_2} = 4.00$$

### **SECTION 3**

- This section contains TWO (02) paragraphs. Based on each paragraph, there are TWO (02) questions.
- Each question has FOUR options. ONLY ONE of these four options corresponds to the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If ONLY the correct option is chosen.

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered).

Negative Marks : -1 In all other cases.

## **PARAGRAPH "X"**

In electromagnetic theory, the electric and magnetic phenomena are related to each other. Therefore, the dimensions of electric and magnetic quantities must also be related to each other. In the questions below, [E] and [B] stand for dimensions of electric and magnetic fields respectively, while  $[\in_0]$  and  $[\mu_0]$  stand for dimensions of the permittivity and permeability of free space respectively. [L] and [T] are dimensions of length and time respectively. All the quantities are given in SI units.

(There are two questions based on PARAGRAPH "X", the question given below is one of them)

**15.** The relation between [E] and [B] is

$$(A) [E] = [B] [L] [T]$$

(B) [E] = [B] 
$$[L]^{-1}$$
 [T]

(C) [E] = [B] [L] 
$$[T]^{-1}$$

(D) [E] = [B] 
$$[L]^{-1}$$
 [T]<sup>-1</sup>

Ans. [C]

**Sol.** For an electromagnetic wave propagation.

$$C = \frac{E}{B}$$

So 
$$E = B. C$$

$$[E] = [B] [C] [T]^{-1}$$

**16.** The relation between  $[\in_0]$  and  $[\mu_0]$  is

(A) 
$$[\mu_0] = [\epsilon_0] [L]^2 [T]^{-2}$$

(B) 
$$[\mu_0] = [\epsilon_0] [L]^{-2} [T]^2$$

(C) 
$$[\mu_0] = [\epsilon_0]^{-1} [L]^2 [T]^{-2} (D) [\mu_0] = [\epsilon_0]^{-1} [L]^{-2} [T]^2$$

Ans. [D]

Sol. we know that

$$C \,=\, \frac{1}{\sqrt{\mu_0 \,\in_0}}$$

$$C^2 = \frac{1}{\mu_0 \, \in_0}$$

$$\left[\mu_{0}\right]=\left[\epsilon_{0}\right]^{-1}\left[L\right]^{-2}\left[T\right]^{2}$$

### **PARAGRAPH "A"**

If the measurement errors in all the independent quantities are known, then it is possible to determine the error in any dependent quantity. This is done by the use of series expansion and truncating the expansion at the first power of the error. For example, consider the relation z = x/y. If the errors in x, y and z are  $\Delta x$ ,  $\Delta y$  and  $\Delta z$ , respectively, then

$$z \pm \Delta z = \frac{x \pm \Delta x}{y \pm \Delta y} = \frac{x}{y} \left( 1 \pm \frac{\Delta x}{x} \right) \left( 1 \pm \frac{\Delta y}{y} \right)^{-1}$$

The series expansion for  $\left(1\pm\frac{\Delta y}{y}\right)^{-1}$ , to first power in  $\Delta y/y$ , is  $1\mp(\Delta y/y)$ . The relative errors in

independent variables are always added. So the error in z will be

$$\Delta z = z \left( \frac{\Delta x}{x} + \frac{\Delta y}{y} \right).$$

The above derivation makes the assumption that  $\Delta x/x <<1$ ,  $\Delta y/y <<1$ . Therefore, the higher powers of these quantities are neglected.

(Three are two questions based on PARAGRAPH "A", the question given below is one of them)

17. Consider the ratio  $r = \frac{(1-a)}{(1+a)}$  to be determined by measuring a dimensionless quantity a. If the error in

the measurement of a is  $\Delta a$  ( $\Delta a/a \ll 1$ ), then what is the error  $\Delta r$  in determining r?

(A) 
$$\frac{\Delta a}{(1+a)^2}$$

(B) 
$$\frac{2\Delta a}{(1+a)^2}$$

(C) 
$$\frac{2\Delta a}{(1-a^2)}$$

(D) 
$$\frac{2a\Delta a}{(1-a^2)}$$

Ans. [B]

**Sol.** 
$$r = \frac{(1-a)}{(1+a)}$$

$$\frac{\Delta r}{r} = \frac{\Delta (1-a)}{(1-a)} + \frac{\Delta (1+a)}{(1+a)}$$

$$\frac{\Delta r}{r} = \frac{\Delta a}{(1-a)} + \frac{\Delta a}{(1+a)}$$

$$\frac{\Delta r}{r} = \frac{\Delta a(1+a+1-a)}{(1-a)(1+a)}$$

$$\frac{\Delta r}{r} = \frac{2\Delta a}{(1-a)(1+a)}$$

$$\Delta r = \frac{2\Delta a}{(1-a)(1+a)} \times r$$

$$\Delta r = \frac{2\Delta a}{(1-a)(1+a)} \times \frac{(1-a)}{(1+a)}$$

$$\Delta r = \frac{2\Delta a}{(1+a)^2}$$

18. In an experiment the initial number of radioactive nuclei is 3000. It is found that 1000  $\pm$  40 nuclei decayed in the first 1.0 s. For |x| << 1,  $\ln(1 + x) = x$  up to first power in x. The error  $\Delta\lambda$ , in the determination of the decay constant  $\lambda$ , in s<sup>-1</sup>, is

Ans. [C

Sol. From radioactivity

$$N = N_0 e^{-\lambda t}$$

$$\ell nN = \ell n N_0 - \lambda t$$

$$\frac{\Delta N}{N} = \Delta \lambda t$$

$$\Delta \lambda = \frac{40}{2000 \times t} = \frac{4}{200}$$

$$\Delta \lambda = 0.02$$

## **PART B: CHEMISTRY**

- **19.** The compound(s) which generate(s) N<sub>2</sub> gas upon thermal decomposition below 300°C is (are):
  - (A) NH<sub>4</sub>NO<sub>3</sub>

(B)  $(NH_4)_2Cr_2O_7$ 

(C)  $Ba(N_3)_2$ 

(D)  $Mg_3N_2$ 

Ans. [B,C]

**Sol.** (A)  $\stackrel{-3}{N}H_4\stackrel{+5}{N}O_3 \xrightarrow{above 300^{\circ}C} N_2 \uparrow +H_2O \uparrow +O_2 \uparrow$ 

(B)  $(NH_4)_2Cr_2O_7 \xrightarrow{\text{below } 300^{\circ}C} N_2(g) \uparrow + Cr_2O_3(s) + H_2O \uparrow$ 

(C)  $Ba(N_3)_2 \xrightarrow{below 300^{\circ}C} Ba(s) + N_2(g)$ 

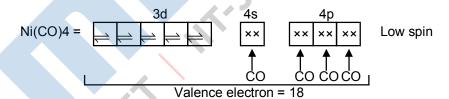
 $(D) Mg_3N_2 \xrightarrow{above 300^{\circ}C} Mg(s) + N_2(g) \uparrow$ 

- 20. The correct statement(s) regarding the binary transition metal carbonyl compounds is (are)

  (Atomic numbers: Fe = 26, Ni = 28)
  - (A) Total number of valence shell electrons at metal centre in Fe(CO)<sub>5</sub> or Ni(CO)<sub>4</sub> is 16
  - (B) These are predominantly low spin in nature
  - (C) Metal-carbon bond strengthens when the oxidation state of the metal is lowered
  - (D) The carbonyl C-O bond weakens when the oxidation state of the metal is increased

Ans. [B,C]

Sol. (A)



(C & D)

as negative charge  $\uparrow$  or as positive charge  $\downarrow$  donation by metal  $\uparrow$  M–C Bond order  $\uparrow$ 

e in ABMO of CO ↑ CO Bond order ↓

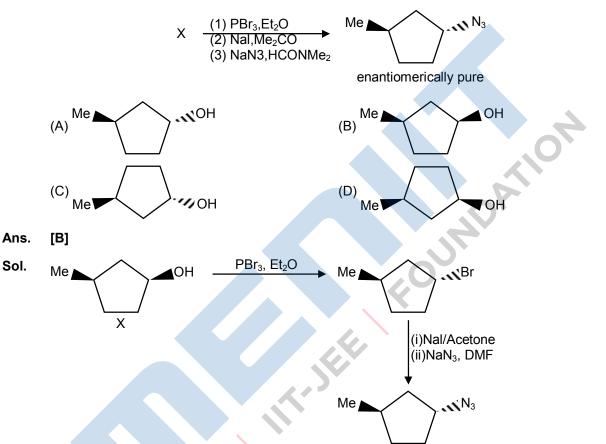
- 21. Based on the compounds of group 15 elements, the correct statement(s) is (are)
  - (A) Bi<sub>2</sub>O<sub>5</sub> is more basic than N<sub>2</sub>O<sub>5</sub>
  - (B) NF<sub>3</sub> is more covalent than BiF<sub>3</sub>
  - (C) PH3 boils at lower temperature than NH3
  - (D) The N-N single bond is stronger than the P-P single bond

Ans. [A, B,C]

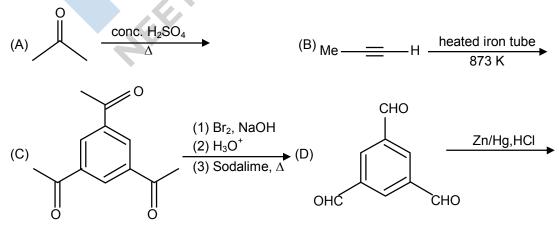
**Sol.** (A)  $N_2^{+5}O_5 > Bi_2O_5 > (Acidic)$ 

Acidic strength  $\alpha$  oxidation state  $\alpha$  Non Metallic character.

- ∴ Bi<sub>2</sub>O<sub>5</sub> is more basic
- (B) NF<sub>3</sub> is more covalent BiF<sub>3</sub>
- (C) Boiling point of NH<sub>3</sub>> PH<sub>3</sub> (H-bond in NH<sub>3</sub>)
- (D) P P single is stronger than N N single bond due to lone pair lone pair repuslion
- 22. In the following reaction sequence, the correct structure(s) of X is (are):



23. The reaction(s) leading to the formation of 1,3,5-trimethylbenzene is (are)



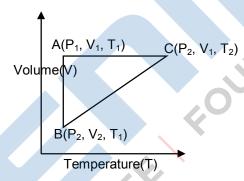
Ans. [A,B,D]

6-1

(A) 
$$\frac{\text{conc.H}_2\text{SO}_4}{\Delta}$$

(Clemmensen reduction)

24. A reversible cyclic process for an ideal gas is shown below. Here, P, V, and T are pressure, volume and temperature, respectively. The thermodynamic parameters q, w, H and U are heat, work, enthalpy and internal energy, respectively



The correct option(s) is (are)

(A) 
$$q_{AC} = \Delta U_{BC}$$
 and  $w_{AB} = P_2(V_2 - V_1)$ 

(B) 
$$W_{BC} = P_2(V_2 - V_1)$$
 and  $q_{BC} = \Delta H_{AC}$ 

(C) 
$$\Delta H_{CA} < \Delta U_{CA}$$
 and  $q_{AC} = \Delta U_{BC}$ 

(D) 
$$q_{BC} = \Delta H_{AC}$$
 and  $\Delta H_{CA} > \Delta U_{CA}$ 

Ans. [B,C]

Sol.

(A) 
$$q_{AC} = \Delta U_{AC} - w_{AC}^{0}$$
$$= C_{V} (T_{2} - T_{1})$$
$$= \Delta U_{BC}$$

but 
$$W_{AB} = 0$$

(B) 
$$W_{BC} = -P_2 (V_1 - V_2) = P_2 (V_2 - V_1)$$
  
 $q_{BC} = C_P (T_2 - T_1)$   
 $= \Delta H_{AC}$ 

(C) 
$$\Delta H_{CA} = C_P (T_1 - T_2) < 0$$
  
 $\Delta U_{CA} = C_V (T_1 - T_2) < 0$ 

$$|\Delta H_{CA}| > |\Delta U_{CA}|$$

but 
$$\Delta H_{CA} < \Delta U_{CA}$$

and from option (A),  $q_{AC} = \Delta U_{BC}$ 

(D) From option (B),  $q_{BC} = \Delta H_{AC}$ but  $\Delta H_{CA} < \Delta U_{CA}$ 

#### **SECTION 2**

25. Among the species given below, the total number of diamagnetic species is \_\_\_\_. H atom,  $NO_2$  monomer,  $O_2^-$  (superoxide), dimeric sulphur in vapour phase,  $Mn_3O_4$ ,  $(NH_4)_2[FeCl_4]$ ,  $(NH_4)_2[NiCl_4]$ ,  $K_2MnO_4$ ,  $K_2CrO_4$ 

Ans. [1]

**Sol.** H-atom Para (1s<sup>1</sup>)

NO<sub>2</sub> (Monomer) Para (odd electron species)

 $O_2^-$  (superoxide) Para (MOT)  $S_2(g)$  Para (MOT)

 $Mn_3O_4$  [ $MnO.Mn_2^{+2}O_3$ ] Para ( $Mn^{2+}$  and  $Mn^{+3}$  both para)

 $(NH_4)[FeCl_4]$  Para  $(d^5, WFL, para)$ 

 $(NH_4)_2[NiCl_4]$  Para  $(d^8, WFL, para)$ 

 $K_2 \stackrel{\text{\tiny +6}}{Mn} O_4$  Para (d<sup>1</sup>)

 $K_2 \overset{+6}{\text{Cr}} O_4$  Dia  $(d^0)$ 

26. The ammonia prepared by treating ammonium sulphate with calcium hydroxide is completely used by NiCl<sub>2</sub>.6H<sub>2</sub>O to form a stable coordination compound. Assume that both the reactions are 100% complete. If 1584 g of ammonium sulphate and 952 g of NiCl<sub>2</sub>.6H<sub>2</sub>O are used in the preparation, the combined weight (in grams) of gypsum and the nickel-ammonia coordination compound thus produced is \_\_\_\_.

(Atomic weights in g mol<sup>-1</sup>: H = 1, N = 14, O = 16, S = 32, CI = 35.5, Ca = 40, Ni = 59)

Ans. [2992]

4 mole

**Sol.**  $(NH_4)_2SO_4 + Ca(OH)_2 \longrightarrow 2NH_3 + CaSO_4.2H_2O$ 

12 mole excess

→24 12

 $NiCl_2.6H_2O + 6NH_3 \longrightarrow Ni(NH_3)_6Cl_2 + 6H_2O$ 

– – 4 mole

24 mole

 $W_{gypsum} + W_{complex} = 12(172) + 4(232) = 2992 grams$ 

- 27. Consider an ionic solid MX with NaCl structure. Construct a new structure (Z) whose unit cell is constructed from the unit cell of MX following the sequential instructions given below. Neglect the charge balance.
  - (i) Remove all the anions (X) except the central one
  - (ii) Replace all the face centered cations (M) by anions (X)
  - (iii) Remove all the corner cations (M)
  - (iv) Replace the central anion (X) with cation (M)

The value of 
$$\left(\frac{\text{number of anions}}{\text{number of cations}}\right)$$
 in **Z** is \_\_\_\_\_.

[3] Ans.

In MX, from the information given, cation M forms FCC structure Sol.

Cation  $M \rightarrow at$  all corners as well as at all face-centers of the unit cell.

Anion  $X \rightarrow$  at all edge centers and at body center of the unit cell.

OUNDATIO on applying all alterations, the structure Z would have as given below:

|              | M | X   |
|--------------|---|-----|
| Initially    | 4 | 4   |
| After step-1 | 4 | 1   |
| After step-2 | 1 | 1+0 |
| After step-3 | 0 | 4   |
| After step-4 | 1 | 3   |

 $\therefore$  Cation M  $\rightarrow$  Only at body center (Z = 1) And Anion  $X \rightarrow At$  all face centers (Z = 3)

$$So\left(\frac{Z_{anion}}{Z_{cation}}\right) = \frac{3}{1} = 3$$

For the electrochemical cell, 28.

the standard emf of the cell is 2.70 V at 300 K. When the concentration of  ${\rm Mg}^{2^+}$  is changed to  ${\bf x}$   ${\bf M}$ , the cell potential changes to 2.67 V at 300 K. The value of **x** is \_\_\_\_\_

(given,  $\frac{F}{R}$  = 11500 K V<sup>-1</sup>, where F is the Faraday constant and R is the gas constant,  $\ln(10)$  = 2.30)

Ans. [10]

Sol. The cell reaction is

$$Mg(s) + Cu^{2+}(a) \rightarrow Mg^{2+}(a) + Cu(s)$$

$$\therefore E_{cell} = E_{cell}^{\circ} - \frac{RT}{nF} ln \frac{[Mg^{2+}]}{[Cu^{2+}]}$$

$$\Rightarrow 2.67 \ = \ 2.70 - \frac{T}{n \left(\frac{F}{R}\right)} ln \frac{[Mg^{2+}]}{[Cu^{2+}]}$$

$$\Rightarrow -0.03 = -\frac{300}{2 \times 11500} ln \left(\frac{x}{1}\right)$$

- $\Rightarrow$  2.3 = Inx
- $\Rightarrow$  In 10 = Inx
- $\therefore x = 10$
- 29. A closed tank has two compartments **A** and **B**, both filled with oxygen (assumed to be ideal gas). The partition separating the two compartments is fixed and is a perfect heat insulator (Figure 1). If the old partition is replaced by a new partition which can slide and conduct heat but does NOT allow the gas to leak across (Figure 2), the volume (in m<sup>3</sup>) of the compartment **A** after the system attains equilibrium is

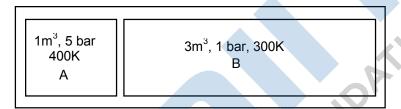


Figure 1

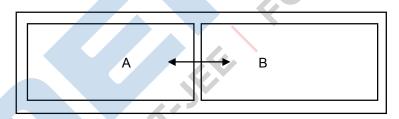


Figure 2

Ans.

P and T will be same in both compartment Sol.

P and T will be same in both comp 
$$\frac{V_1}{V_2} = \frac{n_1}{n_2} = \frac{\frac{P_1 V_1}{T_1}}{\frac{P_2 V_2}{T_2}} = \frac{5 \times 1}{400} \times \frac{300}{3 \times 1} = \frac{5}{4}$$

$$V_A = \frac{5}{9} \times 4 = \frac{20}{9} = 2.22$$

30. Liquids A and B form ideal solution over the entire range of composition. At temperature T, equimolar binary solution of liquids A and B has vapour pressure 45 Torr. At the same temperature, a new solution of  ${\bf A}$  and  ${\bf B}$  having mole fractions  ${\bf x}_{_{\! A}}$  and  ${\bf x}_{_{\! B}}$  respectively, has vapour pressure of 22.5 Torr. The value of  $x_{\Delta}/x_{R}$  in the new solution is \_\_\_\_

(**Given** that the vapour pressure of pure liquid **A** is 20 Torr at temperature T)

Ans. [19]

 $45 = P_A^{\circ} \times \frac{1}{2} + P_B^{\circ} \times \frac{1}{2}$ Sol.

$$P_A^{\circ} + P_B^{\circ} = 90$$

$$P_A^{\circ} + P_B^{\circ} = 90$$
  $\therefore P_B = 70 \text{ Torr}$   $P_A = 20 \text{ Torr}$ 

$$P_{\Delta} = 20 \text{ Torm}$$

$$22.5 = 20 \times x_{\Delta} + 70 (1 - x_{\Delta})$$

$$22.5 = 20 x_A + 70 - 70x_A$$

$$50x_{A} = 70 - 22.5 = 47.5$$

$$x_A = 0.95, x_B = 0.05$$

$$\frac{x_A}{x_B} = \frac{0.95}{0.05} = 19$$

The solubility of a salt of weak acid (AB) at pH 3 is  $Y \times 10^{-3}$  mol L<sup>-1</sup>. The value of Y is 31.

(Given that the value of solubility product of AB ( $K_{sp}$ ) =  $2 \times 10^{-10}$  and the value of ionization constant of COUND WILL **HB** (K<sub>2</sub>) =  $1 \times 10^{-8}$ )

[4.47] Ans.

**Sol.** 
$$AB(s) \rightleftharpoons A^{+}(a) + B^{-}(a)$$

$$K_{sp} = S[B^{-}]$$
....(i)

$$(S-X)$$

$$\frac{1}{K_a} = \frac{[BH]}{[B^-] \times 10^{-3}}$$

$$\Rightarrow 10^8 = \frac{[BH]}{[B^-] \times 10^{-3}}$$

$$\Rightarrow$$
 [BH] = [B $^{-}$ ] × 10 $^{5}$ 

But, by mass balance:

$$S = [B^{-}] + [BH]$$

$$\Rightarrow S = [B^{-}] + [B^{-}] \times 10^{5}$$

$$\Rightarrow$$
 S = [B] (1+10<sup>5</sup>) [B] × 10<sup>5</sup>

$$\Rightarrow \left[ B^{-} \right] = \frac{S}{10^{5}}$$

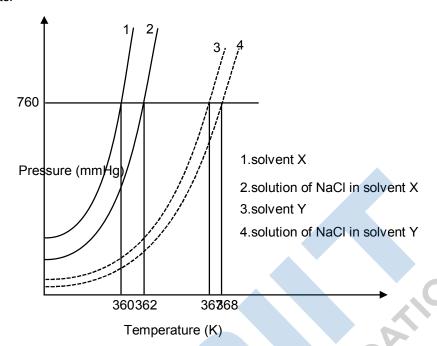
Putting in eq<sup>n</sup>. (i)

$$\Rightarrow 2 \times 10^{-10} = \frac{s^2}{10^5} :: S = \sqrt{20} \times 10^{-3}$$
$$= 4.47 \times 10^{-3}$$

i.e. Solubility of AB = 
$$4.47 \times 10^{-3}$$
 mol/L

$$= Y \times 10^{-3} \text{ mol/L}$$

32. The plot given below shows P-T curves (where P is the pressure and T is the temperature) for two solvents **X** and **Y** and isomolal solutions of NaCl in these solvents. NaCl completely dissociates in both the solvents.



On addition of equal number of moles of a non-volatile solute **S** in equal amount (in kg) of these solvents, the elevation of boiling point of solvent **X** is three times that of solvent **Y**. Solute **S** is known to undergo dimerization in these solvents. If the degree of dimerization is 0.7 in solvent **Y**, the degree of dimerization in solvent **X** is \_\_\_\_\_.

Ans. [0.05]

$$Sol. \qquad \frac{2}{1} = \frac{2 \times K_{b_1} \times m}{2 \times K_{b_2} \times m}$$

$$\frac{2}{1} = \frac{K_{b_1}}{K_{b_2}}$$

$$\Delta T_{b_1} \,=\, 3\,\Delta T_{b_2}$$

$$i_1K_{b_1}m = i_2K_{b_2}m$$

$$i_1 \times 2 = 3i_2$$

$$2A \longrightarrow A_2$$

$$1 - \alpha$$
  $i_1K_{b_1}m = i_2K_{b_2}m$   $i = \frac{1 - \alpha + \frac{\alpha}{2}}{1}$ 

$$i_2 = 1 - \frac{\alpha_2}{2}$$

$$= 1 - 0.35 = 0.65$$

$$i_1 = 3 \times \frac{0.65}{2} = 0.325 \times 3$$

$$1 - \frac{\alpha_1}{2} = 0.325 \times 3$$

$$1 - \frac{\alpha_1}{2} = 0.975$$

$$\alpha_{1} = 0.05$$

## **SECTION 3**

#### Passage "X"

Treatment of benzene with CO/HCl in the presence of anhydrous  $AlCl_3/CuCl$  followed by reaction with  $Ac_2O/NaOAc$  gives compound **X** as the major product. Compound **X** upon reaction with  $Br_2/Na_2CO_3$ , followed by heating at 473 K with moist KOH furnishes **Y** as the major product. Reaction of **X** with  $H_2/Pd-C$ , followed by  $H_3PO_4$  treatment gives **Z** as the major product.

(There are two questions based on PARAGRAPH "X", the question given below is one of them)

СООН

#### **33.** The Compound **Y**is:

Ans. [C]

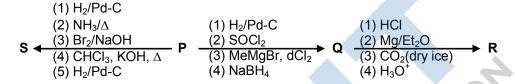
**34.** The Compound **Z**is:

Ans. [A]

Sol. (A) 
$$(B)$$
  $(C)$   $(D)$ 

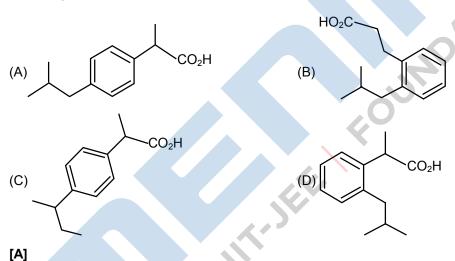
#### Passage "A"

An organic acid  $\mathbf{P}$  ( $C_{11}H_{12}O_2$ ) can easily be oxidized to a dibasic acid which reacts with ethyleneglycol to produce a polymer dacron. Upon ozonolysis,  $\mathbf{P}$  gives an aliphatic ketone as one of the products.  $\mathbf{P}$  undergoes the following reaction sequences to furnish  $\mathbf{R}$  via The compound  $\mathbf{P}$  also undergoes another set of reactions to produce  $\mathbf{S}$ .



## **35.** The compound **R**is:

Ans.



Sol.

#### **36.** The compound **S** is:

Ans. [B]

Sol. 
$$H_2/N_1$$
  $H_2/N_2$   $H_2/N_3$   $H_3/A$   $H$ 

## PART B: MATHEMATICS

#### **SECTION 1**

- 37. For a non-zero complex number z, let arg(z) denote the principal argument with  $-\pi < \arg(z) \le \pi$ . Then, which of the following statement(s) is(are) **FALSE**?
  - (A) arg  $\left(-1 i\right) = \frac{\pi}{4}$ , where  $i = \sqrt{-1}$ .
  - (B) The function  $f: R \to (-\pi, \pi]$ , defined by f(t) = arg(-1 + it) for all  $t \in R$ , is continuous at all points of R, where i =  $\sqrt{-1}$ .
  - (C) For any two non-zero complex numbers  $z_1$  and  $z_2$ ,  $arg\left(\frac{z_1}{z_2}\right)$   $arg\left(z_1\right)$  +  $arg\left(z_2\right)$  is an integer multiple of  $2\pi$ .
  - (D) For any three given distinct complex numbers  $z_1$ ,  $z_2$  and  $z_3$ , the locus of the point z satisfying the OUNDATIC condition  $\arg\left(\frac{(z-z_1)(z_2-z_3)}{(z-z_3)(z_2-z_4)}\right) = \pi$  lies on a straight line.

[ABD] Ans.

**Sol.** (a) 
$$\arg \left(-1 - i\right) = -\pi + \frac{\pi}{4} = \frac{-3\pi}{4}$$

(b) 
$$f(t) = arg(-1 + it) = \pi - tan^{-1}(t)$$
  $t > 0$   
 $= \pi$   $t = 0$   
 $= -\pi + tan^{-1} |t|$   $t < 0$ 

Discontinuous at t = 0

(c) 
$$\arg\left(\frac{z_1}{z_2}\right) = \arg(z_1) - \arg(z_2) + 2k\pi$$

(d) 
$$= \frac{(z - z_1)(z_2 - z_3)}{(z - z_3)(z_2 - z_1)} = \frac{(\overline{z} - \overline{z}_1)(\overline{z}_2 - \overline{z}_3)}{(\overline{z} - \overline{z}_3)(\overline{z}_2 - \overline{z}_1)}$$

It represent straight line as well as circle for different values of  $z_1$ ,  $z_2$  and  $z_3$ .

- 38. In a triangle PQR, let  $\angle$ PQR = 30° and the sides PQ and QR have lengths 10  $\sqrt{3}$  and 10, respectively. Then, which of the following statement(s) is (are) TRUE?
  - (A)  $\angle$ QPR = 45°.
  - (B) The area of the triangle PQR is  $25\sqrt{3}$  and  $\angle$ QRP =  $120^{\circ}$ .
  - (C) The radius of the incircle of the triangle PQR is  $10 \sqrt{3} 15$ .
  - (D) The area of the circumcircle of the triangle PQR is  $100\pi$ .

[BCD] Ans.

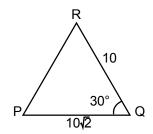
**Sol.** 
$$\cos 30^{\circ} = \frac{10^2 + (10\sqrt{3})^2 - x^2}{2 \cdot 10 \cdot 10\sqrt{3}} \frac{\sqrt{3}}{2}$$

$$\Rightarrow$$
 x = 10 = PR

Ar (
$$\triangle$$
PQR) =  $\frac{1}{2}$  (PR) (QR) · sin 120° =  $25\sqrt{3}$ 

$$r = \frac{\Delta}{S} = \frac{25\sqrt{3}}{10 + 5\sqrt{3}} = \frac{5\sqrt{3}}{2 + \sqrt{3}} = 5\sqrt{3}(2 - \sqrt{3})$$

$$R = \frac{PQ}{2\sin 120^{\circ}} = 10$$



- Let  $P_1$ : 2x + y z = 3 and  $P_2$  = x + 2y + z = 2 be two planes. Then, which of the following statement(s) 39. is (are) TRUE?
  - (A) The line of intersection of  $P_1$  and  $P_2$  has direction ratios 1, 2, –1.
  - (B) The line  $\frac{3x-4}{9} = \frac{1-3y}{9} = \frac{z}{3}$  is perpendicular to the line of intersection of P<sub>1</sub> and P<sub>2</sub>.
  - (C) The acute angle between  $P_1$  and  $P_2$  is  $60^\circ$ .
  - (D) If  $P_3$  is the plane passing through the point (4, 2, -2) and perpendicular to the line of intersection of  $P_1$  and  $P_2$ , then the distance of the point (2, 1, 1) from the plane  $P_3$  is  $\frac{2}{\sqrt{3}}$

[CD] Ans.

Sol.

Line of intersection
$$\vec{n}_1 \times \vec{n}_2 = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 2 & 1 & -1 \\ 1 & 2 & 1 \end{vmatrix} = 3\hat{i} - 3\hat{j} + 3\hat{k}$$

Line  $\frac{x-\frac{4}{3}}{3} = \frac{y-\frac{1}{3}}{-3} = \frac{Z}{3}$  is parallel to line of intersection.

Angle between plane  $P_2$  and  $P_2$ 

$$cos\theta = \frac{2+2-1}{\sqrt{6} \cdot \sqrt{6}} = \frac{1}{2} \Rightarrow \theta = 60^{\circ}$$

Equation of plane P<sub>3</sub> is

$$1(x-4)-(y-2)+(z+2)=0$$

$$\Rightarrow$$
  $x - y + z = 0$ 

Perpendicular distance from (2, 1, 1) is  $\frac{2}{\sqrt{3}}$ .

- For every twice differentiable function f : R  $\rightarrow$  [- 2, 2] with  $\left(f(0)\right)^2+\left(f'(0)\right)^2$  = 85, which of the 40. following statement(s) is (are) TRUE?
  - (A) There exist  $r, s \in R$ , where r < s, such that f is one-one on the open interval (r, s).
  - (B) There exist  $x_0 \in (-4, 0)$  such that  $|f'(x_0)| \le 1$ .
  - (C)  $\lim_{x \to \infty} f(x) = 1$
  - (D) There exists  $\alpha \in (-4, 4)$  such that  $f(\alpha) + f''(\alpha) = 0$  and  $f'(\alpha) \neq 0$ .

Ans. [ABD]

 $f: R \rightarrow [-2, 2]$ Sol.

$$(f(0))^2 + (f'(0))^2 = 85$$

$$(f(0))^2 \le 4$$
  $\Rightarrow (f'(0))^2 \ge 81$ 

y = f (x) is not a constant function and we know that if a function is not constant, then it must be one-one in (r, s) for some  $r, s \in R$ .

According to LMVT

$$f'(x_0) = \frac{f(0) - f(-4)}{0 - (-4)} = \frac{f(0) - f(-4)}{4}$$

$$\Rightarrow \qquad \left| f'(x_0) \right| = \left| \frac{f(0) - f(-4)}{4} \right| \le \left| \frac{4}{4} \right| \le 1 \qquad \text{For at least one } x_0 \in (-4, 0) \qquad \dots (1)$$

Similarly

$$|f'(x_1)| \le 1$$
 For at least one  $x_1 \in (0, 4)$  .....(2)

 $f'(0) \ge 81$ and

 $g(x) = f^{2}(x) + (f'(x))^{2}$  be a function, then g(x) must be differentiable. Let

$$g(x_0) \le 5$$
;  $g(0) = 85$ ;  $g(x_1) \le 5$ 

There must be at least one 'a' and 'b'

$$a \in (-4, 0)$$
 and  $b \in (0, 4)$  such that  $g(a) = g(b)$ 

- g'(x) = 0 has a real root in (a, b)
- f(x) + f''(x) = 0 has real root.
- Let  $f: R \to R$  and  $g: R \to R$  be two non-constant differentiable functions. If  $f'(x) = (e^{(f(x)-g(x))})g'(x)$ 41. for all  $x \in \mathbb{R}$ , and f(1) = g(2) = 1, then which of the following statement(s) is (are) **TRUE**?

(A) 
$$f(2) < 1 - \log_{2} 2$$

(B) 
$$f(2) > 1 - log 2$$

(B) 
$$f(2) > 1 - \log_e 2$$
 (C)  $g(1) > 1 - \log_e 2$  (D)  $g(1) < 1 - \log_e 2$ 

(D) 
$$a(1) < 1 - log 2$$

Ans.

**Sol.** 
$$\int e^{-f(x)} \cdot f'(x) dx = \int e^{-g(x)} \cdot g'(x) dx$$
$$-e^{-f(x)} = -e^{-g(x)} + c$$

$$-e^{-g(x)}-e^{-f(x)}=c$$

Case 1: 
$$e^{-g(x)} - e^{-f(x)} = e^{-g(1)} - \frac{1}{e} = C$$

$$e^{-g(2)} - e^{-f(2)} = e^{-g(1)} - \frac{1}{e}$$

$$e^{-f(2)} = \frac{2}{e} - e^{-g(1)}$$

$$e^{-f(2)}<\frac{2}{e}$$

$$f(2) > 1 - \ell n 2$$

**Case 2:** 
$$e^{-g(x)} - e^{-f(x)} = c$$

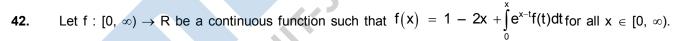
$$e^{-g(2)} - e^{-f(2)} = c = e^{-g(x)} - e^{-f(x)}$$

$$\frac{1}{e} - e^{-f(2)} = e^{-g(x)} - e^{-f(x)}$$

$$\frac{2}{e} - e^{-f(2)} = e^{-g(1)}$$

$$e^{-g(1)} = \frac{2}{e} - e^{-f(2)} < \frac{2}{e}$$

$$g(1) > 1 - \ell n 2$$



Then, which of the following statement(s) is(are) TRUE?

- (A) The curve y = f(x) passes through the point (1, 2).
- (B) The curve y = f(x) passes through the point (2, -1).

(C) The area of the region 
$$\left\{ (x,y) \in [0,1] \times R : f(x) \le y \le \sqrt{1-x^2} \right\}$$
 is  $.\frac{\pi-2}{4}$ 

(D) The area of the region 
$$\left\{ \ (x,y) \in [0,1] \times R : f(x) \leq y \leq \sqrt{1-x^2} \ \right\}$$
 is  $\frac{\pi-1}{4}$ .

Ans. [BC]

Sol. 
$$f(x) = 1 - 2x + e^x \int_0^x e^{-t} \cdot f(t) dt$$

$$f'(x) = -2 + e^{x} \int_{0}^{x} e^{-t} \cdot f(t) \cdot dt + e^{x} \cdot e^{-x} \cdot f(x)$$

$$\frac{dy}{dx} = -2 + (y + 2x - 1) + y$$

$$\frac{dy}{dx} - 2y = 2x - 3$$

$$y \times e^{-2x} = \int e^{-2x} (2x - 3) dx$$

$$y \times e^{-2x} = \frac{e^{-2x}}{-2}(2x-3) + \int e^{-2x} \cdot dx$$

$$=\frac{e^{-2x}}{-2}(2x-3)-\frac{e^{-2x}}{2}+C$$

$$2v = -(2x - 3) - 1 + (2c) \cdot e^{2x}$$

(: f(0) = 1, then c = 0)

$$y = -x + 1$$

Ar = 
$$\frac{\pi}{4} - \frac{1}{2}$$

#### **SECTION 2**

- The value of  $\left( (\log_2 9)^2 \right)^{\frac{1}{\log_2(\log_2 9)}} \times \left( \sqrt{7} \right)^{\frac{1}{\log_4 7}}$  is 43.
- Ans. [8.00]
- Sol.  $4 \cdot 2 = 8$
- The number of 5digit numbers which are divisible by 4, with digits from the set {1, 2, 3, 4, 5} and the 44. repetition of digits is allowed, is
- [625.00] Ans.

$$\underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{$$

Sol. 
$$--\frac{3}{2} \xrightarrow{2} 125$$
  
 $--\frac{4}{4} \xrightarrow{4} 125$ 

$$\underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} \underline{\phantom{a}} 125$$

- Let X be the set consisting of the first 2018 terms of the arithmetic progression 1, 6, 11, ......, and 45. Y be the set consisting of the first 2018 terms of the arithmetic progression 9, 16, 23, ...... Then, the number of elements in the set  $X \cup Y$  is \_\_\_\_\_.
- [3748.00] Ans.

**Sol.** 
$$n(X \cup Y) = n(X) + n(Y) - n(X \cap Y)$$
  
= 2018 + 2018 - 288 = 3748

46. The number of real solution of the equation

$$\sin^{-1}\left(\sum_{i=1}^{\infty} x^{i+1} - x \sum_{i=1}^{\infty} \left(\frac{x}{2}\right)^{i}\right) = \frac{\pi}{2} - \cos^{-1}\left(\sum_{i=1}^{\infty} \left(\frac{-x}{2}\right)^{i} - \sum_{i=1}^{\infty} (-x)^{i}\right) = \text{lying in the interval}$$

$$\left(\frac{-1}{2},\frac{1}{2}\right)$$
 is \_\_\_\_\_.



[Here, the inverse trigonometric functions  $\sin^{-1}x$  and  $\cos^{-1}x$  assume values in  $\left[\frac{-\pi}{2}, \frac{\pi}{2}\right]$  and  $[0, \pi]$ , respectively.]

Ans. [2.00]

**Sol.** = 
$$\sin^{-1} \left( \frac{x^2}{1-x} - \frac{x(\frac{x}{2})}{1-\frac{x}{2}} \right) + \cos^{-1} \left( \frac{(-x)}{2} - \frac{(-x)}{1+x} \right) = \frac{\pi}{2}$$

$$\sin^{-1}\left(\frac{x^2}{1-x} - \frac{x^2}{2-x}\right) + \cos^{-1}\left(\frac{x}{1+x} - \frac{x}{x+2}\right) = \frac{\pi}{2}$$

$$\sin^{-1}\left(\frac{x^2}{(x-1)(x-2)}\right) + \cos^{-1}\left(\frac{x}{(x+1)(x+2)}\right) = \frac{\pi}{2}$$

$$\Rightarrow \qquad \frac{x^2}{(x-1)(x-2)} = \frac{x}{(x+1)(x+2)}$$

$$\Rightarrow x = 0 \text{ or } x(x^2 + 3x + 2) = x^2 - 3x + 2$$
$$x^3 + 2x^2 + 5x - 2 = 0$$

- 47. For each positive integer n, let  $y_n = \frac{1}{n} ((n+1)(n+2).....(n+n))^{\frac{1}{n}}$ . For  $x \in R$ , let [x] be the greatest integer less than or equal to x. If  $\lim_{n\to\infty} y_n = L$ , then the value of [L] is \_\_\_\_\_.
- Ans. [1.00]

**Sol.** 
$$y_n = \left( \left( 1 + \frac{1}{n} \right) \left( 1 + \frac{2}{n} \right) \left( 1 + \frac{3}{n} \right) .... \left( 1 + \frac{n}{n} \right) \right)^{\frac{1}{n}}$$

$$\underset{n \to \infty}{\text{Lim}} \, y_n = \underset{n \to \infty}{\text{Lim}} \Bigg( \Bigg( 1 + \frac{1}{n} \Bigg) \Bigg( 1 + \frac{2}{n} \Bigg) \dots \Bigg( 1 + \frac{n}{n} \Bigg) \Bigg)^{\frac{1}{n}} = \underset{n \to \infty}{\text{Lim}} \, e^{\frac{1}{n} \sum_{r=1}^n \ell n \left( 1 + \frac{r}{n} \right)} = e^{\frac{1}{0} \ell n (1 + x) \cdot dx} \\ = e^{\ell n \cdot 4 - 1} = \frac{4}{e}.$$

- 48. Let  $\vec{a}$  and  $\vec{b}$  be two unit vectors such that  $\vec{a} \cdot \vec{b} = 0$ . For some  $x, y \in R$ , let  $\vec{a} \cdot \vec{b} = 0$ . If  $|\vec{c}| = 2$  and the vector  $\vec{c}$  is inclined at the same angle  $\alpha$  to both  $\vec{a}$  and  $\vec{b}$ , then the value of  $8\cos^2 \alpha$  is \_\_\_\_\_.
- Ans. [3.00]

**Sol.** 
$$\overline{a} \cdot \overline{c} = x = 2 \cos \alpha$$

$$\overline{b} \cdot \overline{c} = y = 2 \cos \alpha$$

$$\overline{c} = (2\cos\alpha)\overline{a} + (2\cos\alpha)\overline{b} + (\overline{a} \times \overline{b})$$

$$|\bar{c}| = 4 \cos^2 \alpha + 4 \cos^2 \alpha + 1 = 4 \implies 8 \cos^2 \alpha = 3$$

49. Let a, b, c be three non-zero real numbers such that the equation  $\sqrt{3}$  a cos x + 2b sin x = c, x  $\in \left[\frac{-\pi}{2}, \frac{\pi}{2}\right]$ ,

has two distinct real roots  $\alpha$  and  $\beta$  with  $\alpha + \beta = \frac{\pi}{3}$ . Then, the value of  $\frac{b}{a}$  is \_\_\_\_\_.

**Sol.** 
$$\sqrt{3}a \frac{\left(1-\tan^2\frac{x}{2}\right)}{\left(1+\tan^2\frac{x}{2}\right)} + 2b \left(\frac{2\tan\frac{x}{2}}{1+\tan^2\frac{x}{2}}\right) = c$$

$$\sqrt{3}a\left(1-tan^2\frac{x}{2}\right)+4b\left(tan\frac{x}{2}\right)=c\left(1+tan^2\frac{x}{2}\right)$$

$$\left(c+\sqrt{3}a\right)tan^{2}\frac{x}{2}-4btan\frac{x}{2}+\left(c-\sqrt{3}a\right)=0<\frac{tan\frac{\alpha}{2}}{tan\frac{\beta}{2}}$$

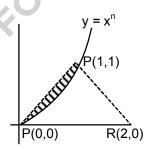
$$tan\left(\frac{\alpha}{2} + \frac{\beta}{2}\right) = tan\frac{\pi}{6} = \frac{1}{\sqrt{3}} = \frac{\left(\frac{4b}{c + \sqrt{3}a}\right)}{1 - \left(\frac{c - \sqrt{3}a}{c + \sqrt{3}a}\right)} =$$

$$\Rightarrow \frac{1}{\sqrt{3}} = \frac{4b}{2\sqrt{3}a} \Rightarrow \frac{b}{a} = \frac{1}{2} = 0.50. \text{ Ans.}$$

- 50. A farmer  $F_1$  has a land in the shape of a triangle with vertices at P(0, 0), Q (1, 1) and R (2, 0). From this land, a neighbouring farmer  $F_2$  takes away the region which lies between the side PQ and a curve of the form  $y = x^n$  (n > 1). If the area of the region taken away by the farmer  $F_2$  is exactly 30% of the area of  $\Delta$ PQR, then the value of n is \_\_\_\_\_\_.
- Ans. [4.00]
- **Sol.** Area of shaded region =  $\frac{3}{10}$  (Ar. ( $\triangle PQR$ ))

$$\frac{1}{2} - \int_{0}^{1} x^{n} dx = \frac{3}{10} \left( \frac{1}{2} \cdot 2 \cdot 1 \right)$$

$$\frac{1}{2} - \frac{1}{n+1} = \frac{3}{10} \Rightarrow \frac{1}{n+1} = \frac{1}{5} \Rightarrow n = 4.$$
 Ans.



#### **SECTION 3**

# Paragraph for Question no. 15 & 16

Let S be the circle in the xy-plane defined by the equation  $x^2 + y^2 = 4$ .

51. Let  $E_1$   $E_2$  and  $F_1F_2$  be the chords of S passing through the point  $P_0$  (1, 1) and parallel to the x-axis and the y-axis, respectively. Let  $G_1$   $G_2$  be the chord of S passing through  $P_0$  and having slope – 1. Let the tangents to S at  $E_1$  and  $E_2$  meet at  $E_3$ , the tangents to S at  $E_1$  and  $E_2$  meet at  $E_3$ , the points  $E_3$ ,  $E_3$  and  $E_4$  and  $E_5$  meet at  $E_5$ . Then, the points  $E_3$ ,  $E_5$  and  $E_5$  are on the curve

$$(A) x + y = 4$$

(B) 
$$(x-4)^2 + (y-4)^2 = 16$$

(C) 
$$(x-4)(y-4)=4$$

(D) 
$$xy = 4$$

Ans. [A]

**52.** Let P be a point on the circle S with both coordinates being positive. Let the tangent to S at P intersect the coordinate axes at the points M and N. Then, the mid-point of the line segment MN must lie on the curve

(A) 
$$(x + y)^2 = 3xy$$

(B) 
$$x^{\frac{2}{3}} + y^{\frac{2}{3}} = 2^{\frac{4}{3}}$$

(C) 
$$x^2 + y^2 = 2xy$$

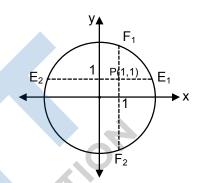
(D) 
$$x^2 + y^2 = x^2 y^2$$

Ans. [D]

Sol.

(i) Let point  $E_3$  ( $x_1$ ,  $y_1$ ) such that  $xx_1 + yy_1 = 4$  is same as y = 1  $\Rightarrow x_1 = 0, y_1 = 4$ Let point  $F_3$  ( $x_2$ ,  $y_2$ ) such that  $xx_2 + yy_2 = 4$  is same as x = 1  $\Rightarrow x_2 = 4 \text{ and } y_2 = 0$ 

Let point  $G_3(x_3, y_3)$  such that  $xx_3 + yy_3 = 4$  is same as x + y = 2 $\Rightarrow x_3 = y_3 = 2$ 



(ii) Let equation of line MN is  $\frac{x}{2h} + \frac{y}{2k} = 1$ 

$$\frac{1}{\sqrt{\frac{1}{4h^2} + \frac{1}{4k^2}}} = 2 \Rightarrow \frac{1}{x^2} + \frac{1}{y^2} = 1.$$

## Paragraph for Question no. 17 & 18

There are five students  $S_1$ ,  $S_2$ ,  $S_3$ ,  $S_4$  and  $S_5$  in a music class and for them there are five seats  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$  and  $R_5$  arranged in a row, where initially the seat  $R_i$  is allotted to the student  $S_i$ , i = 1, 2, 3, 4, 5. But, on the examination day, the five students are randomly allotted the five seats.

- 53. The probability that, on the examination day, the student  $S_1$  get the previously allotted seat  $R_1$ , and NONE of the remaining students gets the seat previously allotted to him/her is
  - (A)  $\frac{3}{40}$
- (B)  $\frac{3}{40}$
- (C)  $\frac{7}{40}$
- (D)  $\frac{1}{5}$

Ans. [A]

- **54.** For i = 1, 2, 3, 4, let  $T_i$  denote the event that the student  $S_i$  and  $S_{i+1}$  do **NOT** sit adjacent to each other on the day of the examination. Then, the probability of the event  $T_1 \cap T_2 \cap T_3 \cap T_4$  is
  - (A)  $\frac{1}{15}$
- (B)  $\frac{1}{10}$
- $(C)\frac{7}{60}$
- (D)  $\frac{1}{5}$

Ans. [C]

Sol.

(i) 
$$D_4 = 4! \left( \frac{1}{2!} - \frac{1}{3!} + \frac{1}{4!} \right) = 12 - 4 + 1 = 9$$

$$P(E) = \frac{9}{5!} = \frac{9}{120} = \frac{3}{40}.$$

(ii) 
$$S_1 \stackrel{S_3}{\smile} S_5 \stackrel{S_2}{\smile} S_4$$
  $\Rightarrow$  2 ways

$$\Rightarrow$$
 3 ways

$$S_2$$
 $S_4$ 
 $S_1$ 
 $S_5$ 
 $S_5$ 
 $S_5$ 
 $S_3$ 
 $S_5$ 
 $S_5$ 
 $S_3$ 
 $S_5$ 
 $S_5$ 

$$S_4 \rightarrow 3$$
 ways as in  $S_2$ 

$$\mathrm{S_5}\!\to\!2$$
 ways as in  $\mathrm{S_1}$ 

Hence required probability =  $\frac{14}{5!} = \frac{7}{60}$  Ans.